

NCERT/CBSE CHEMISTRY CLASS 11 textbook

<http://www.TutorBreeze.com>
English, French, Accounts, MBA

Contact for Online Tutoring in Physics, Math, Chemistry,

Solutions/Answers to NCERT/CBSE CHEMISTRY Class 11(Class XI)textbook

CHAPTER TWELVE

ORGANIC CHEMISTRY – SOME BASIC PRINCIPLES AND TECHNIQUES

12.31 Why is it necessary to use acetic acid and not sulphuric acid for acidification of sodium extract for testing sulphur by lead acetate test?

It necessary to use acetic acid and not sulphuric acid for acidification of sodium extract for testing sulphur by lead acetate test because sulphuric acid will react with lead acetate

To form a precipitate of lead sulphate, thus interfering in the test of sulphur.

